

## Cycloaddition Reactions of Carbodiimides. The First Example of an Intramolecular Diels–Alder Reaction of C=C-Conjugated Carbodiimides

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An one-pot preparation of  $\alpha$ -carbolines and quinindolines from conjugated carbodiimides based on a tandem intramolecular Diels–Alder cycloaddition/oxidative aromatization is described.

In recent years there has been an upsurge of interest in the [4 + 2] cycloaddition of azadienes.<sup>1</sup> In spite of the important role of the intramolecular [4 + 2] cycloaddition of 2-azadienes, only one example of this reaction, involving electron-poor 2-azadienes and unactivated dienophiles, has been reported.<sup>2</sup> On the other hand, a number of heterocumulenes have been shown to function as 2-azadienes in [4 + 2] cycloaddition reactions, *e.g.* *N*-aryl ketenimines,<sup>3</sup> *N*-aryl vinylketenimines,<sup>4</sup> phenyl isocyanate and related isocyanates<sup>5</sup> react with elec-

tron-rich dienophiles by cycloaddition across the 2-azadiene system; however there have been no reports dealing with unsaturated carbodiimides as 2-azadienes, to the best of our knowledge. Continuing our interest on the preparation and synthetic applications of C=C-conjugated carbodiimides,<sup>6</sup> we report herein a one-pot preparation of indole derivatives, based on the strategy shown in Scheme 1. Our synthetic approach, which involves as the key step a tandem intramolecular hetero Diels–Alder cycloaddition–oxidative aromatiza-

tion has surprisingly been found to be useful in the simultaneous formation of pyrrole and pyridine rings in the synthesis of *[b]*-fused pyridoindoles.

Recently, pyrido[2,3-*b*]indoles ( $\alpha$ -carbolines) (**4**) and quinindolines (**9**) have received much attention because they are pharmacologically active compounds, displaying strong cytostatic and antitumour activity.<sup>7</sup>

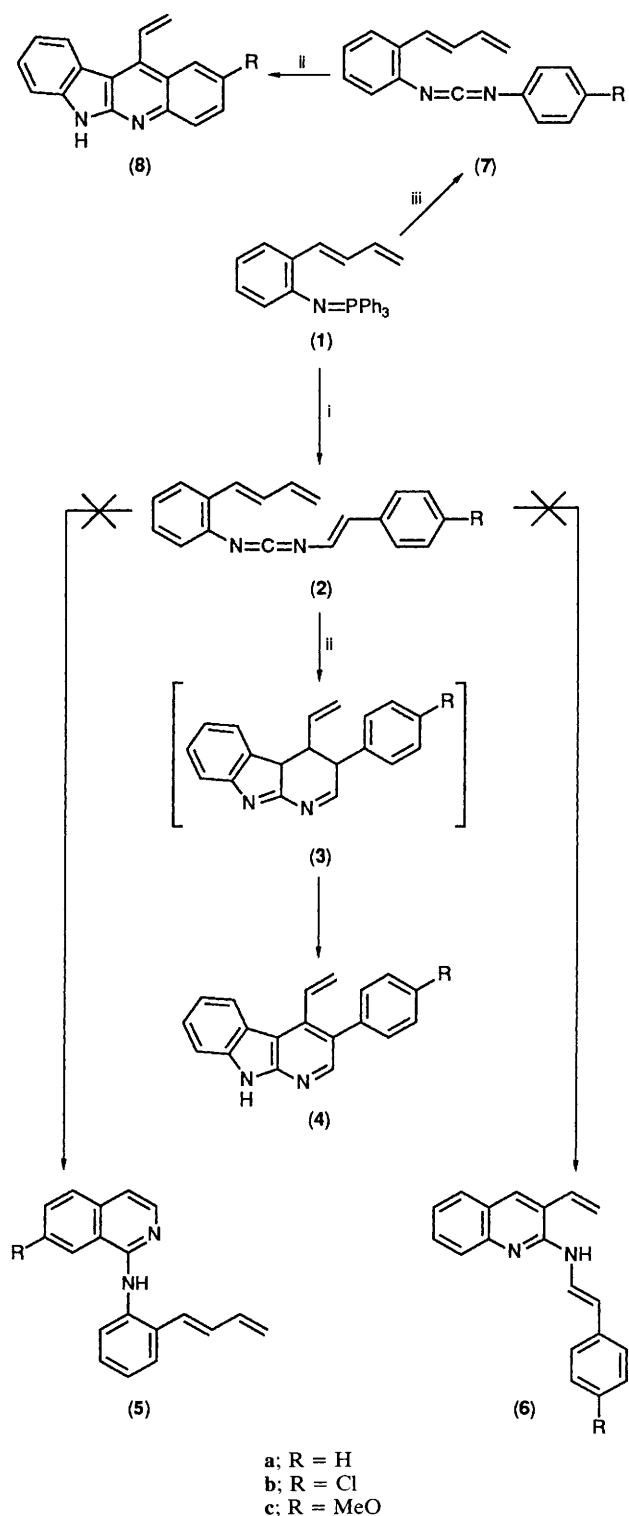
The key intermediate (**1**) was easily prepared in 47% overall yield from 1-(*o*-nitrophenyl)buta-1,3-diene<sup>8</sup> (1:1 mixture of *E/Z* isomers) by standard chemistry: iodine-catalysed isomerization to the *E* isomer, reduction with the iron-acetic acid system and reaction with triphenylphosphine dibromide in the presence of triethylamine. Aza-Wittig reaction of the iminophosphorane (**1**) with styryl isocyanates in dry toluene at room temperature led to the corresponding carbodiimides (**2**) bearing a diene chain in one *ortho* position, which could be isolated as viscous oils by means of short column chromatography. When toluene solutions of (**2**) were heated at 160 °C for 16 h the corresponding tricyclic compounds (**4**) were obtained as solids, in moderate yields, after column chromatography (silica gel, ethyl acetate-*n*-hexane 1:1) accompanied by minor amounts of unidentified compounds in which neither the isoquinoline (**5**) nor quinoline (**6**) could be detected.

In the light of the behaviour of the previously reported conjugated carbodiimides,<sup>6</sup> the intramolecular reactivity of the carbodiimides (**2**) came as a surprise. The former compounds underwent electrocyclic ring-closure followed by 1,3-proton shift to give either isoquinoline or quinoline derivatives.

In an analogous reaction sequence, the diarylcarbodiimides (**7**), available from (**1**) and aromatic isocyanates, were converted into the corresponding tetracyclic compounds (**8**) which were isolated as crystalline solids. That conversions of the type (**2**)  $\rightarrow$  (**4**) and (**7**)  $\rightarrow$  (**8**) are reasonably general in nature is indicated by the examples given in Table 1. Efforts to improve the yields of (**4**) and (**8**) by heating the carbodiimides (**2**) and (**7**) in the presence of palladium-charcoal as a dehydrogenation agent were unsuccessful. In spite of the moderate yields, the syntheses of (**4**) and (**8**)† are competitive with known routes to comparable compounds, and take place in a completely periselective fashion.

These transformations include a [4 + 2] cycloaddition whereby the unsaturated carbodiimide has functioned as a

2-azadiene using one cumulative double bond and a C=C double bond adjacent to the cumulative system, whereas the C=C double bond of the *ortho*-butadienyl substituent directly linked to the aromatic ring has taken the role of the dienophile. A final oxidative aromatization followed by a



**Scheme 1.** Reagents and conditions: i, ArCH=CHNCO, toluene, room temp.; ii, sealed tube, toluene, 160 °C, 16 h; iii, ArNCO, toluene, room temp.

† Selected spectroscopic data for 3-(4-methoxyphenyl)-4-vinyl-9*H*-pyrido[2,3-*b*]indole (**4c**): <sup>1</sup>H NMR (200 MHz, [<sup>2</sup>H<sub>6</sub>] DMSO) δ 3.80 (s, 3H, MeO), 5.55 (dd, 1H, *J* 1.7 and 17.9 Hz, CH=CH<sub>2</sub>), 5.73 (dd, 1H, *J* 1.7 and 11.5 Hz, CH=CH<sub>2</sub>), 7.00 [d, 2H, *J* 8.5 Hz, H(3') and H(5')], 7.06–7.22 [m, 2H, CH=CH<sub>2</sub> and H(6)], 7.33 [d, 2H, *J* 8.6 Hz, H(2') and H(6')], 7.42–7.58 [m, 2H, H(7') and H(8')], 8.25 [d, 1H, *J* 8.1 Hz, H(5'')], 8.29 [s, 1H, H(2)], and 11.94 (s, 1H, NH); <sup>13</sup>C NMR (50 MHz, [<sup>2</sup>H<sub>6</sub>] DMSO) δ 55.02 (MeO), 111.31 [C(8)], 111.69 [C(3)], 113.63 [C(3' and 5')], 119.14 [C(6)], 120.34 [C(4a)], 122.62 (CH=CH<sub>2</sub>), 122.77 [C(5)], 126.31 [C(7)], 127.16 [C(4b)], 130.35 [C(1')], 131.32 [C(2' and 6')], 133.05 (CH=CH<sub>2</sub>), 139.33 [C(8a)], 139.45 [C(4)], 146.83 [C(2)], 151.48 [C(9a)], and 158.24 [C(4')]. 2-Methoxy-11-vinyl-6*H*-quinindoline (**8c**): <sup>1</sup>H NMR (200 MHz, [<sup>2</sup>H<sub>6</sub>] DMSO) δ 3.90 (s, 3H, MeO), 5.95 (dd, 1H, *J* 1.7 and 17.8 Hz, CH=CH<sub>2</sub>), 6.13 (dd, 1H, *J* 1.7 and 11.6 Hz, CH=CH<sub>2</sub>), 7.16–7.67 (m, 6H, aromatic and CH=CH<sub>2</sub>), 7.95 [d, 1H, *J* 9.1 Hz, H(4)], 8.28 [d, 1H, *J* 7.8 Hz, H(10)], and 11.68 (s, 1H, NH); <sup>13</sup>C NMR (50 MHz, [<sup>2</sup>H<sub>6</sub>] DMSO) δ 55.12 (MeO), 103.26 [C(1)], 110.67 [C(7)], 114.07 [C(11a)], 119.04 [C(9)], 120.25 [C(10b)], 120.71 [C(3)], 122.18 [C(10a)], 123.38 [C(10)], 123.48 (CH=CH<sub>2</sub>), 127.59 [C(8)], 128.73 [C(4)], 131.95 (CH=CH<sub>2</sub>), 137.32 [C(11)], 141.55 [C(6a)], 142.15 [C(4a)], 151.38 [C(5a)], and 154.68 [C(2)]. Values assigned by decoupling methods and 2D <sup>1</sup>H-<sup>13</sup>C correlation techniques. DMSO = dimethyl sulphoxide.

**Table 1.** Yields of pyrido[2,3-*b*]indoles (**4**) and quinindolines (**8**).<sup>a</sup>

	R	Yield (%)
( <b>4a</b> )	H	37
( <b>4b</b> )	Cl	46
( <b>4c</b> )	MeO	35
( <b>8a</b> )	H	39
( <b>8b</b> )	Cl	43
( <b>8c</b> )	MeO	40

<sup>a</sup> All new compounds described here had spectral and microanalytical properties in agreement with the assigned structures.

1,3-proton shift in the cycloadduct furnishes the polycycle (**4**) or (**8**) respectively.‡

In conclusion, this work shows for the first time that easily available C=C-conjugated carbodiimides may react as 2-azadienes in intramolecular [4 + 2] cycloadditions; obviously the structural conditions in *N*-aryl(styryl), *N'*-(*o*-butadienyl)-phenyl carbodiimides provide an energetically favourable situation for this exceptional behaviour. It can be presumed that related carbodiimides may also undergo this type of

‡ Carbodiimide type (**7**) derived from (**1**) and 2,6-dimethylphenyl isocyanate does not undergo cycloaddition; it was recovered unaltered after prolonged heating (toluene, 160 °C, 4 days). Obviously the two methyl groups at the *ortho* positions prevent the cycloaddition step.

cycloaddition to give nitrogen heterocycles related to the antitumour agent ellipticine.

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## References

- 1 D. L. Boger, *Tetrahedron*, 1983, **39**, 2869; D. L. Boger and S. N. Weinreb, in 'Hetero Diels-Alder Methodology in Organic Synthesis,' Academic Press, San Diego, 1987.
- 2 J. Barluenga, M. Tomás, A. Ballesteros, and V. Gotor, *J. Chem. Soc., Chem. Commun.*, 1989, 267.
- 3 L. Ghosez and C. de Pérez, *Angew. Chem., Int. Ed. Engl.*, 1971, **10**, 184.
- 4 E. Souveaux and L. Ghosez, *J. Am. Chem. Soc.*, 1973, **95**, 5117.
- 5 M. E. Kuehne and P. J. Sheeran, *J. Org. Chem.*, 1968, **33**, 4406.
- 6 P. Molina, P. M. Fresneda, and P. Alarcón, *Tetrahedron Lett.*, 1988, **29**, 379; P. Molina, A. Arques, P. M. Fresneda, M. V. Vinader, M. C. Foces-Foces, and F. H. Cano, *Chem. Ber.*, 1989, **122**, 307; P. Molina, M. Alajarín, and A. Vidal, *J. Chem. Soc., Chem. Commun.*, 1990, 7.
- 7 W. Peczynska-Czoch, *Arch. Immunol. Ther. Exp.*, 1987, **35**, 221; L. Kaczmarek, R. Balicki, P. Nautka-Namirski, W. Peczynska-Czoch, and M. Mordarski, *Arch. Pharm. (Weinheim, Ger.)*, 1988, **321**, 463.
- 8 E. A. Brander and J. S. Fawcett, *J. Chem. Soc. C*, 1951, 3113.